

BOARD OF INTERMEDIATE AND SECONDARY EDUCATION, MULTAN.
OBJECTIVE KEY FOR INTERMEDIATE 1ST ANNUAL EXAMINATION, 2025

Name of Subject: Chemistry (Home Economics Group) Group: H-Eco Paper I

Q. Nos	Paper Code	Correct Answer	Paper Code	Correct Answer	Paper Code	Correct Answer	Paper Code	Correct Answer
1	C	Cooling						
2	A	Electron						
3	B	Hydrocarbons						
4	D	HNO ₃						
5	A	H ₂ O ₂						
6	C	Tin						
7	B	CO						
8	A	water + salt						
9	D	All of these						
10	C	Both A and B						
11								
12								
13								
14								
15								
16								
17								
18								
19								
20								
Cross Check	A	B	C	D	A	B	C	D
Total	3	2	3	2				

ہر ٹیکٹ بابت صحیح سوال پرچہ امارتنگ Key

ہم نے مضمون کیمسٹری (Home Economics Group) کو Subjctive & Objective (Subjective & Objective) کے دو طریقوں میں چیک کر لیا ہے۔ یہ پرچہ Syllabus کے تین مطابق Set کیا گیا ہے۔ اس سوالیہ پرچہ میں کسی قسم کی کوئی غلطی نہ ہے۔ ہم نے سوالیہ پرچہ کا اردو اور انگریزی Version بھی چیک کر لیا ہے۔ یہ Version آپس میں مطابقت رکھتے ہیں۔ نیز اس پرچہ کی Key (MCQs) کی بابت تصدیق کی جاتی ہے کہ اس میں بھی کسی قسم کی کوئی غلطی نہ ہے۔ مزید یہ کہ ہم نے Key بنانے سے متعلق دفتر کی جانب سے تیار کردہ ہدایات اور اس کے ان پوائنٹس کو مدعا کر لیا ہے اور ان کی روشنی میں Key بنائی ہے۔ نیز سب ایگزامینرز کیلئے تفصیلی مارکنگ ہدایات / مارکنگ سکیم / Rubrics بھی تیار کر دی گئی ہیں۔

Prepared & Checked By:

Dated: _____

S.#	Name	Designation	Institution	Mobile No	Signature
1	Sadia Hanif	Lecturer	Gout Graduate C of H-Eco	03216391104	
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3					
4					
5					

2025 (1 st -A)		Roll No: _____
INTERMEDIATE PART-I (11 th Class)		
CHEMISTRY (HOME ECONOMICS GROUP) PAPER-I		
TIME ALLOWED: 1.15 Hours	SUBJECTIVE	MAXIMUM MARKS: 25
NOTE: Write same question number and its parts number on answer book, as given in the question paper.		
SECTION-I		
2. Attempt any six parts.		6 × 2 = 12
(i)	What is Protein?	
(ii)	Write some properties of air.	
(iii)	What is hard water?	
(iv)	Differentiate between Solute and Solvent.	
(v)	State Law of Mass Action.	
(vi)	What are Reversible Reactions?	
(vii)	Write some properties of Bases.	
(viii)	Give some uses of Acids.	
(ix)	Differentiate between Strong and Weak acids.	
SECTION-II		
NOTE: Attempt any two questions.		2 × 6½ = 13
3.	Define carbohydrate and write down their functions.	
4.	Write a note on fermentation.	
5.	Define atomic structure and write down Dalton's atomic theory.	

Paper Code Number: 2741		2025 (1 st -A) INTERMEDIATE PART-I (11 th Class)		Roll No: _____	
CHEMISTRY (HOME ECONOMICS GROUP) PAPER-I					
TIME ALLOWED: 15 Minutes		OBJECTIVE		MAXIMUM MARKS: 10	
Q.No.1	You have four choices for each objective type question as A, B, C and D. The choice which you think is correct, fill that bubble in front of that question number, on bubble sheet. Use marker or pen to fill the bubbles. Cutting or filling two or more bubbles will result in zero mark in that question.				
S.#	QUESTIONS	A	B	C	D
1	Evaporation causes _____.	Heating	Freezing	Cooling	Melting
2	The outermost shell of atom contains:	Electron	Proton	Neutron	All of these
3	What is principal constituent of petroleum and natural gas?	Carboxylic Acid	Hydrocarbon	Alcohol	Esters
4	Formula for Nitric Acid is:	H ₂ O ₂	CH ₃ COOH	HCl	HNO ₃
5	_____ is used as an antiseptic.	H ₂ O ₂	NaCl	HNO ₃	CH ₃ COOH
6	In tinning process which metal is:	Lead	Copper	Tin	Gold
7	Which gas is known as silent killer?	O ₂	CO	NH ₃	CH ₄
8	What is the end product of neutralization reaction?	Salt + Water	Water	Salt	None of these
9	Metals are _____	Malleable	Ductile	Shiny	All of these
10	Hard water has ion of _____.	Calcium	Magnesium	A and B Both	Sulphur

91(Obj)-2025(1st-A)-50 (MULTAN)